Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, et DOCKET NO.: 040302/0259

1/8

FIG.1A

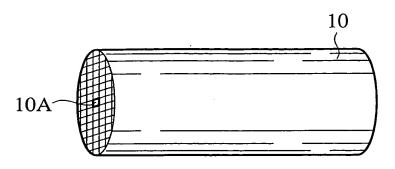


FIG.1B

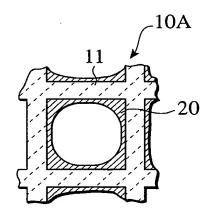
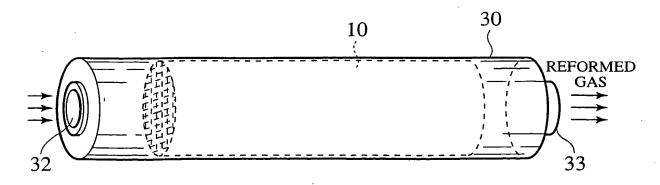
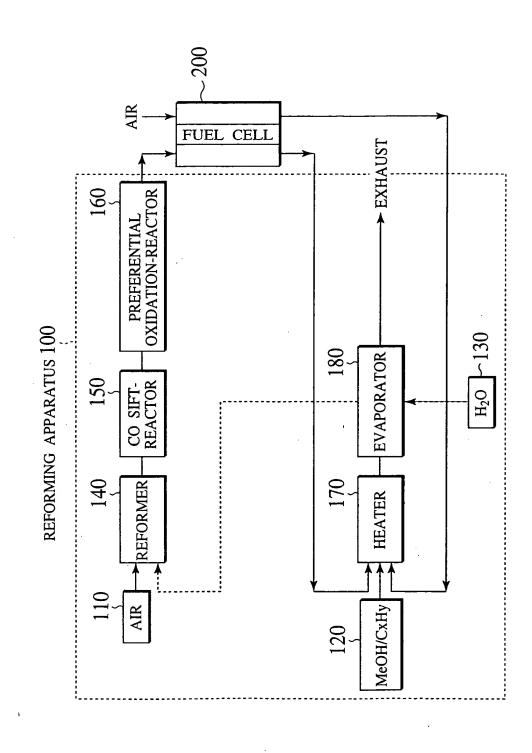


FIG.2



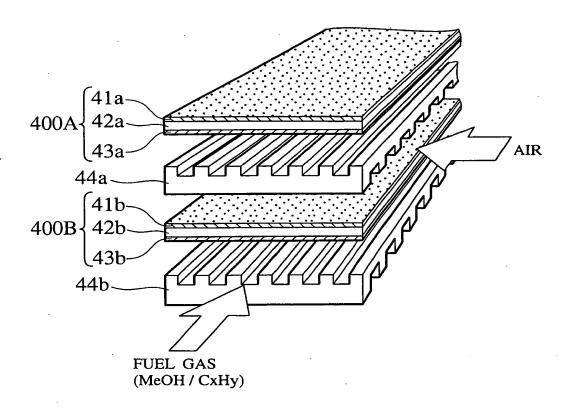
Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, et al DOCKET NO.: 040302/0259

2/8



3/8

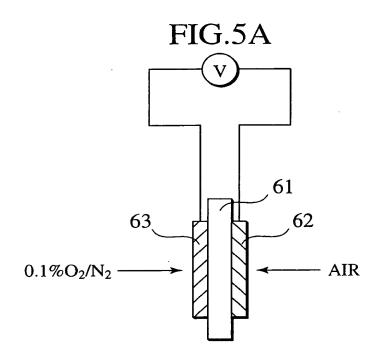
FIG.4

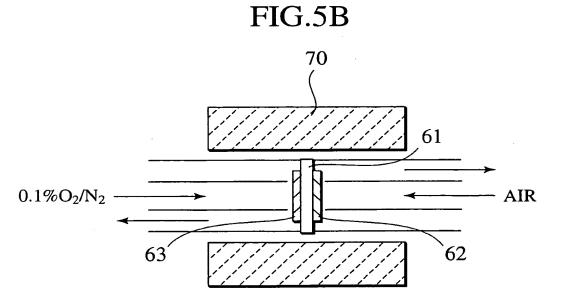


A' H. Hall I Hade A' Ben, H. H.

Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, e DOCKET NO.: 040302/0259

4/8





Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, et al DOCKET NO.: 040302/0259

FIG. 6

Table.1

						5	/8												
Pt-nitrate (8.5wt% solution)		,	,	•		•	J			-	1	•		•	1	1	1835	•	1
Pd-nitrate (8.5wt% solution)	,	1	1	•	1	t	,	1	,	1	,	ı		,		1127	_	ı	-
Rh-nitrate (8.5wt% solution)	•	-	ı	240	240	240	909	896	1	ı	1	1	1	1	240	ı	-	ı	ı
Ru-nitrate (3.6wt% solution)	140.4	140.4	140.4	-		•	ı	1	561.7	561.7	561.7	1404	2247	561.7	•	•	1	140.4	,
Fe-nitrate (g)	•	•	-	323.2	323.2	323.2	202	80.8	323.2	323.2	323.2	707	8.08	-	•	8.08	8.08	,	1
Sm-nitrate Fe-nitrate (g)	44.4	8.88	44.4	-	-	٠	,	•		•	-	•	-	6.88	6.88	-	•	44.4	44.4
Ce-nitrate (g)	-	•	390.6	-	-	t	•	1	1	•	1	1	1	347.2	347.2	•	1	ı	3
Gd-nitrate · (g)	428.5	428.5	428.5	1	•	-	,	t		-	-	-	-	360.8	360.8	-	1	428.5	428.5
Pr-nitrate (g)	•	-	•	43.5	87.0	130.5	87.0	0.78	43.5	0.78	130.5	0.78	0.78	-	-	0.78	87.0	•	1
La-nitrate (g)	389.7	346.4	-	389.7	346.4	303.1	346.4	346.4	389.7	346.4	303.1	346.4	346.4	•	.1	346.4	346.4	389.7	389.7
Example No.	example 1	example 2	example 3	example 4	example 5	example 6	example 7	example 8	example 9	example 10	example 11	example 12	example 13	example 14	example 15	example 16	example 17	Comparative example 1	Comparative example 2

Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO et al DOCKET NO.: 040302/02

6/8

Table.2				FIG.7	7					-	
				Amount	of eacl	n eleme	nt per	Amount of each element per a catalyst unit	st unit		
Example No.	Catalyst	La	Pr	gg	ප	Sm	Fe	Ru	Rh	Pd	P
		(g)	(g)	(g)	(g)	(g)	(g)	(g)	(g)	(g)	<u>8</u>
example 1	Y	4.25	1	5.19	1	0.52	_	0.15	-	'	'
example 2	В	3.74	-	5.19	•	1.03	_	0.15	-	-	•
example 3	S	,	-	5.19	4.08	1.03	_	0.15	-	-	'
example 4	D	6.16	69.0	•	•	-	181	-	96.0	_	•
example 5	Ε	5.47	1.38	•	•		1.81	-	96.0	-	•
example 6	F	4.79	2.07	-	-	1	181	-	96.0	-	•
example 7	ß	5.47	1.38	1	ı	-	1.07	ı	2.29	-	1
example 8	Н	5.47	1.38	-	,	,	0.39	•	3.39	•	, :
example 9	I	6.16	69.0	-	ı	•	1.81	06.0	1	-	,
example 10	ſ	5.47	1.38	_	-	•	1.81	0.00	•	_	'
example 11	K	4.79	2.07	1	-		1.81	06:0	-	-	,
example 12	Г	5.47	1.38	-	•	•	1.07	2.11	-	-	'
example 13	M	5.47	1.38	-	-	ı	0.39	3.14	-	-	ı
example 14	Z	-	-	4.49	3.72	1.06	•	0.62	-	_	'
example 15	0	•	-	4.49	3.72	1.06	1	-	0.70	-	•
example 16	Ь	5.47	1.38	-	-		0.39	١	-	2.88	'
example 17	ð	5.47	1.38	-	١	,	0.39	,	-	-	4.6
Comparative example 1	Х	4.25	-	5.19	-	0.52	-	0.15	1	•	'
Comparative example 2	Y	4.25	-	5.19	1	0.52	ı	•	•	-	. •

Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, al DOCKET NO.: 040302/02

7/8

FIG.8

Table.3

Example No.	Catalyst	Catalyst composition	Amount of remaining MeOH (%)	Reformation rate (%)
example 1	A	La _{0.9} Sm _{0.1} Gd _{0.95} Ru _{0.05} O ₃	0.98	99.05
example 2	В	La _{0.8} Sm _{0.2} Gd _{0.95} Ru _{0.05} O ₃	96:0	99.04
example 3	C	Ce _{0.9} Sm _{0.1} Gd _{0.95} Ru _{0.05} O ₃	1.02	86.86
example 4	Q	La _{0.9} Pr _{0.1} Fe _{0.8} Rh _{0.2} O ₃	0.54	99.46
example 5	3		0.48	99.52
example 6	Ы	$Rh_{0.2}$	0.40	09.66
example 7	Ð	La _{0.8} Pr _{0.2} Fe _{0.5} Rh _{0.5} O ₃	0.37	99.63
example 8	H	La _{0.8} Pr _{0.2} Fe _{0.2} Rh _{0.8} O ₃	0.23	99.77
example 9	I	La _{0.9} Pr _{0.1} Fe _{0.8} Ru _{0.2} O ₃	0.72	99.28
example 10	ſ	La _{0.8} Pr _{0.2} Fe _{0.8} Ru _{0.2} O ₃	0.68	99.32
example 11	Ж	La _{0.7} Pr _{0.3} Fe _{0.8} Ru _{0.2} O ₃	0.51	99.49
example 12	Т	La _{0.8} Pr _{0.2} Fe _{0.5} Ru _{0.5} O ₃	0.46	99.54
example 13	M	La _{0.8} Pr _{0.2} Fe _{0.2} Ru _{0.8} O ₃	0.38	99.62
example 14	N	Ce _{0.8} Sm _{0.2} Gd _{0.8} Ru _{0.2} O ₃	0.65	99.35
example 15	0	Ce _{0.8} Sm _{0.2} Gd _{0.8} Rh _{0.2} O ₃	0.53	99.47
example 16	ď	La _{0.8} Pr _{0.2} Fe _{0.2} Pd _{0.8} O ₃	0.37	99.63
example 17	Ò	La _{0.8} Pr _{0.2} Fe _{0.2} Pt _{0.8} O ₃	0.40	09.66
Comparative example 1	X	La _{0.9} Sm _{0.1} Gd _{0.95} O ₃ / Ru*	2.62	97.38
Comparative example 2	Y	La _{0.9} Sm _{0.1} Gd _{0.95} O ₃	3.64	96.36

* The Ru is impregnated into the perovskite composite oxide.



Title: CATALYST COMPOSITION Inventor(s): Hiroaki KANEKO, et DOCKET NO.: 040302/0259

8/8

FIG.9

Table.4

Example No.	Composition of electrode-catalyst	Electrode- catalyst	Temperature of starting operation Tne (°C)
example 18	La _{0.8} Pr _{0.2} Fe _{0.8} Rh _{0.2} O ₃	a	420
example 19	La _{0.8} Pr _{0.2} Fe _{0.5} Rh _{0.5} O ₃	b	405
example 20	La _{0.8} Pr _{0.2} Fe _{0.2} Rh _{0.8} O ₃	С	387
Comparative example 3	La _{0.8} Pr _{0.2} Fe _{0.8} O ₃	z	650